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 $\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$
 how many grams

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 $\text{CO} + \text{O}_2 \rightarrow \text{CO}_2$
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 NaCl 58.44 g/mol
 $\frac{13.8 \text{ g}}{58.44 \text{ g/mol}} = 0.236 \text{ mol}$
 text g end text of sodium chloride express the answer using 3 significant figures your answer should be
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