

# Características De Los Gases Nobles

**CLARAMUNT VALLESPÍ Rosa M<sup>a</sup> , CORNAGO RAMÍREZ Pilar , ESTEBAN SANTOS Soledad , FARRÁN MORALES Angeles , PÉREZ TORRALBA Marta , SANZ DEL CASTILLO Dionisia**

Química inorgánica Therald Moeller, 1981-02

*Química: la Ciencia Central* Theodore L. Brown, H. Eugene LeMay, Bruce Edward Bursten, Julia R. Burdge, 2004 The scientific accuracy, clarity, and visuals of Chemistry: The Central Science make it the most trusted chemistry book available. A comprehensive media package works in tandem with the text. Translated into Spanish. Introduction: Matter and Measurement, Atoms, Molecules, and Ions, Stoichiometry: Calculations with Chemical Formulas and Equations, Aqueous Reactions and Solution Stoichiometry, Thermochemistry, Electronic Structure of Atoms, Periodic Properties of the Elements, Basic Concepts of Chemical Bonding, Molecular Geometry and Bonding Theories, Gases, Intermolecular Forces, Liquids, and Solids, Modern Materials, Properties of Solutions, Chemical Kinetics, Chemical Equilibrium, Acid-Base Equilibria, Additional Aspects of Aqueous Equilibria, Chemistry of the Environment, Chemical Thermodynamics, Electrochemistry, Nuclear Chemistry, Chemistry of the Nonmetals, Metals and Metallurgy, Chemistry of Coordination Compounds, The Chemistry of Life: Organic and Biological Chemistry. For anyone interested in learning more about chemistry.

Química Andoni Garriz, 1998

Química 1. Fundamentos ,

**Química** , 2005 CONTENIDO: Introducción a la química - Sistema de medidas - Materia y energía - La estructura del átomo - Clasificación periódica de los elementos - Estructura de los compuestos - Nomenclatura química de los compuestos inorgánicos - Cálculos que comprenden elementos y compuestos - Ecuaciones químicas - Cálculo en las ecuaciones químicas. Estequiometría - Gases - Líquidos y sólidos - Agua - Disoluciones y coloides - Ácidos, bases y ecuaciones iónicas - Ecuaciones de oxidación-reducción y electroquímica - Velocidades de reacción y equilibrio químico - Química orgánica - Química nuclear.

**Fundamentos de la química general e inorgánica** Hans Rudolf Christen, 1977 Este texto tiene como objetivo proporcionar a los estudiantes una introducción en los fundamentos teóricos de nuestra ciencia que les permita alcanzar simultáneamente una primera versión general sobre el campo de la Química inorgánica. De acuerdo con esta idea, la parte general -teoría atómica y de enlaces, termodinámica, cinética, reacciones ácido-base y redox, Química de los complejos-

ocupa un espacio bastante amplio, mientras que la Química descriptiva -campo del estudio de las sustancias propiamente dicho- queda más bien en un segundo término sin formar el objetivo fundamental de la obra.

**Química** G. William Daub, William S. Seese, 1996

**Química general** Hans Rudolf Christen, 1975 El libro Química general, escrito como libro de enseñanza para centros de bachillerato, escuelas de comercio, seminarios y otras instituciones docentes de nivel medio se publicó por primera vez en Suiza, en 1963 y, desde entonces se han efectuado seis ediciones sin cambios notables. Como corresponde a su título, este libro da en primer lugar una visión y conocimiento generales de la materia. Porque no es precisamente un conocimiento detallado y específico de la materia lo que da un valor a la enseñanza de la Química, sino un conocimiento del conjunto de leyes fundamentales y de su aplicación al mundo material, de las relaciones principales en el mismo, así como de los métodos (y limitaciones!) de la investigación.

**introducción a la química descriptiva ,**

Química 2 ,

*Química. Fundamentos experimentales* Robert W. Parry, 1973 Este libro es una versión autorizada del célebre libro original CHEM Study, cuya versión española se ha difundido por los Centros de Enseñanza donde se cursan estudios de nivel medio y preuniversitario. Tanto en este libro, como en el CHEM, la experimentación es el vehículo para presentar la Química, tal como es hoy en día. Los principios unificadores se desarrollan a partir de la observación experimental. Así, la Química aparece como una Ciencia y no como una masa de información. El objetivo principal de este libro es lograr fundamentalmente el hito de la Ciencia moderna: desarrollo del principio a partir de la observación.

Química general Ramírez Regalado, Víctor Manuel, 2020-05-13 Química general. 2ª. Ed. Esta obra se diseñó con base en los programas de estudio actualizados de la Escuela Nacional Preparatoria de la UNAM. Esta materia abarca un curso anual y básico para el estudio de la química. La obra incluye los contenidos conceptuales, procedimentales y actitudinales que marca la ENP-UNAM. Toda la información está totalmente actualizada y el libro cuenta con una gran variedad de recursos didácticos.

*Química. La ciencia básica* DOMINGUEZ REBOIRAS, MIGUEL ANGEL, 2006-01-01 Esta obra se ha escrito pensando en las necesidades de los estudiantes de los primeros cursos de Facultades de Ciencias, Farmacia, Ciencias Ambientales y de la Salud, y Escuelas Técnicas.

Química General Víctor Manuel Ramírez Regalado, 2000 Química General, del profesor Ramírez Regalado, es una obra realizada bajo un enfoque actualizado de los temas tratados en esta disciplina y diseñada e impresa a todo color. De esta manera, constituye una herramienta de gran utilidad para la enseñanza de la química en el nivel medio superior. Está conformada por cinco unidades que apoyan las explicaciones de los temas en cuadros, tablas, esquemas e ilustraciones a

todo color, así como en actividades experimentales y autoevaluaciones. Así, el estudiante adquirirá diversos elementos que le facilitarán la adquisición de los conocimientos fundamentales para entender la importancia de la química y su relación con nuestro medio ambiente, pudiendo disponer, además, con esta edición, de diferentes recursos académicos en línea.

Fundamentos de Química Ralph A. Burns, 2003

**Unidad 7 - Introducción a la química (Operaciones básicas de laboratorio)** Marian Borja Prado, Ester Orts Fernández, M<sup>a</sup> del Pilar Sánchez Rodríguez, Unidad 7 - Introducción a la química (Operaciones básicas de laboratorio)

**Química 1** Ramírez Regalado, Víctor Manuel, 2020-05-13 Química 1. Materia, enlaces, nomenclatura y reacciones. Segunda edición. Obra que integra temas básicos en la enseñanza de la química. Totalmente actualizado con base en los nuevos programas de la DGETI. Integra los ejes para las ciencias experimentales. Contenido: Eje 1. Relaciona las aportaciones de la ciencia con el desarrollo de la humanidad. Eje 2.. Distingue la estructura y organización de los componentes naturales del planeta. Eje 3. Explica la estructura y organización de los componentes naturales del planeta. Eje 4. Primera parte. Explica el comportamiento e interacción en los sistemas químicos, biológicos, físicos y ecológicos. Eje 2. Segunda parte. Glosario, Bibliografía, Páginas electrónicas.

**Diversificación Ámbito Científico - Tecnológico II - Andalucía 2023** Mercedes Sánchez, Rubén Solís, Ascensión Ruano, 2023 Contenido Investigación científica Sentido numérico La materia Los compuestos químicos Álgebra Funciones Movimiento y fuerzas Energía y Electricidad Sentido estocástico Rocas y minerales Procesos geológicos

**Sistemas químicos. Guía del profesor** C.B.A. (Chemical Bond Approach Project), 1968 La presente Guía ofrece la experiencia de 300 colegas que emplearon material del CBA (algunos durante un período de más de cuatro años), lo que permite ahorrar tiempo y esfuerzo, evitando que se encuentren muchos de los inconvenientes que se presentan al que desarrolla el curso por primera vez.

**PRINCIPALES COMPUESTOS QUÍMICOS CLARAMUNT VALLESPÍ** Rosa M<sup>a</sup> , CORNAGO RAMÍREZ Pilar , ESTEBAN SANTOS Soledad , FARRÁN MORALES Angeles , PÉREZ TORRALBA Marta , SANZ DEL CASTILLO Dionisia , 2015-07-07 La asignatura Principales Compuestos Químicos se incluye dentro del módulo de Formación Básica del Grado en Química. El objetivo general de dicha asignatura es proporcionar un conocimiento básico, por una parte, de la reactividad y propiedades de los compuestos orgánicos y biomoléculas y, por otra, de la química de los elementos de los grupos de la Tabla Periódica y compuestos más importantes a los que dan lugar, así como de las reacciones propias de las sustancias radiactivas. También en formato e-book.

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